

## LOW TEMPERATURE NITRIDATION OF SILICON

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### Field of the Invention

5           This invention relates to the fabrication of integrated circuits on silicon, and specifically to a method of forming a high quality silicon nitride layer at low temperature in an integrated circuit.

### Background of the Invention

          Thermally grown silicon nitride films are rarely used in conventional IC fabrication  
10   because the temperature required to form a silicon nitride film is higher than that for  $\text{SiO}_2$  formation, which higher temperature may result in damage of  $\text{SiO}_2$  layers. Instead, either low-pressure chemical vapor deposition (LPCVD) or plasma-enhanced chemical vapor deposition (PECVD) silicon nitride films are typically used. S.K. Gandhi, *VLSI Fabrication Principles: Silicon and Gallium Arsenide, 2nd ed.*, 1994. Silicon nitride, when deposited near the  
15   stoichiometric  $\text{Si}_3\text{N}_4$  composition, exhibits a high density and excellent diffusion barrier properties. There are instances where a good, high quality thin, *e.g.*, between about one nm and 100 nm, silicon nitride layer is desired. One is the top stack of a silicon oxide gate dielectric, to prevent the diffusion of boron from p-type gates into a channel region. Another is as an etch stop, *e.g.*, under a silicon dioxide layer.

20           The LPCVD technique deposits silicon nitride at a high deposition rate using  $\text{SiH}_4$  and  $\text{NH}_3$  at temperatures of  $700^\circ\text{C}$  to  $900^\circ\text{C}$ . The high deposition rate makes this technique very useful for masking and diffusion blocking applications. The high deposition temperature, however, makes for a high degree of stress in the film when cooled to room temperature. The

PECVD technique also has a high deposition rate at temperatures of between about 275°C to 400°C. Films formed by PECVD have a high quantity of hydrogen incorporated therein and have a higher wet etch rate than a LPCVD film. Another alternative is to reactive sputter silicon nitride with a silicon target and nitrogen in the plasma. Other low temperature methods have been reported, e.g., U.S. Patent No. 6,274,510 B1, for *Lower temperature method for forming high quality silicon nitride dielectrics*, granted August 14, 2001, to Wilk et al.

#### Summary of the Invention

A method of low-temperature nitridation of a silicon substrate includes placing a silicon wafer in a vacuum chamber on a heated chuck; maintaining the silicon wafer at a temperature of between about room temperature and 400 °C; introducing a nitrogen-containing gas into the vacuum chamber; dissociating the nitrogen-containing gas into nitrogen with an excimer lamp and flowing the nitrogen over the silicon wafer; and forming an silicon nitride layer on at least a portion of the silicon wafer.

It is an object of the invention to provide a high quality silicon nitride film for use in an integrated circuit.

Another object of the invention is to provide a silicon nitride film which is formed at reactively low temperature.

This summary and objectives of the invention are provided to enable quick comprehension of the nature of the invention. A more thorough understanding of the invention may be obtained by reference to the following detailed description of the preferred embodiment of the invention in connection with the drawings.

### Brief Description of the Drawings

Fig. 1 is a schematic representation of the apparatus used to achieve the low temperature nitridation.

Fig. 2 depicts the growth rate of the silicon nitride layer at room temperature and 400°C chuck temperature.

Fig. 3 depicts the efficacy of the nitride layer in blocking thermal oxidation.

### Detailed Description of the Preferred Embodiments

The method of the invention disclosed herein employs the use of nitrogen radicals to convert silicon to a silicon nitride. The method of the invention can also form a thin nitride layer on an already-grown silicon oxide layer by displacing the oxygen at the top surface and converting at least a portion of silicon oxide to silicon nitride. Because a silicon nitride layer can be formed at a relatively low temperature, the inherent stress level is significantly reduced. This has been reported to have been done with radicals generated in a plasma, K. Watanabe *et al.*, *Controlling the concentration and position of nitrogen in ultrathin oxynitride films formed by using oxygen and nitrogen radicals*, Appl. Phys. Lett. 76, 2940 (2000). In the method of the invention, it is done without the plasma discharge, which may cause large amounts of silicon damage. The limitation of this technique is that thickness of up to 5 nm is reasonable and thicker films will be nearly impossible to grow unless higher temperatures are employed.

The method of the invention generates large quantities of nitrogen radicals on or near the surface of a silicon layer, or silicon oxide layer which is to be converted to silicon nitride. The radicals are generated by the photolysis of  $\text{NH}_3$ . The light source used here is a  $\text{Xe}_2$  excimer lamp that emits efficiently at a wavelength of 172 nm, or 7.21 eV in energy. The bonding energy

for  $\text{NH}_2\text{-H}$  is 4.8eV, for  $\text{N-H}$  is 3.3eV, and  $\text{NH-H}$  is expected to be somewhere in between these values, so the photon energy is sufficient to cleave the strip the nitrogen in ammonia of it's hydrogen atoms. The ionization potentials of  $\text{NH}_3$ ,  $\text{NH}_2$ ,  $\text{NH}$  and  $\text{N}$  are 10.2eV, 11.4eV, 13.1eV and 14.5eV, respectively, so the formation of ions in the gas phase are improbable.

5           The apparatus used in the method of the invention is depicted in Fig. 1, generally at 10. An excimer lamp 12, which emits light at a wavelength of 172 nm is placed in a vacuum chamber 14 above the surface of silicon wafer 16 that is to be oxidized. Excimer lamp 12 is a xenon-based lamp and is commercially available at a reasonable cost. One such lamp is a Xeradex® lamp produced by Osram Sylvania.

10           A steady flow of a nitrogen-containing gas, such as  $\text{NH}_3$ , is introduced into chamber 14 through inlet 18. The pressure in chamber 14 is controlled by a throttle valve 20, located between the chamber and the pump system. Wafer 16 sits on a heated chuck 22, capable of reaching temperatures of up to about 400°C. The thermal coupling between the wafer and chuck is poor, so the actual wafer temperature may be lower than 250°C when the chuck 15 temperature is set at 400°C. Chamber pressure is controlled to a range between about five mTorr. to 200 mTorr. The flow of  $\text{NH}_3$  is regulated to be between about two sccm and 50 sccm.

Excimer lamp 12 emits at a wavelength of 172 nm. This Xe excimer lamp is commercially available at a reasonable cost. In developing the method of the invention, a Xeradex lamp produced by Osram Sylvania was used. A layer of silicon nitride having a thickness of 20 between six Å and 50 Å may be grown using the method of the invention and apparatus in a time of between about thirty seconds and three hours.

The direct illumination of the wafer surface may generate photoelectrons and a

charged surface that may participate in the nitridation process. The work function of silicon is less than 5eV so electrons can have over 2.2eV of kinetic energy. Electron attachment of the low energy electrons may generate negatively charged species, such as  $\text{NH}_2^-$ , that are quite stable.

Adsorbed molecules on the surface of the substrate may also play a role in the nitride layer growth.

5 The growth of the film may be assisted by a field across the growing dielectric layer where a positively charged interface attracts negative ions, similar to the metal oxidation model proposed by Cabrera and Mott in 1948, as reported by J. Joseph *et al.*, *A kinetics study of the electron cyclotron resonance plasma oxidation of silicon*, J. Vac. Sci. Technol. B10, 611 (1992).

The film growth rates observed do not follow a parabolic function as predicted by  
10 Cabrera-Mott, nor by thermal oxidation models. Fig. 2 shows the observed growth rates in the chamber at room temperature, nominally 15°C, and at a chuck temperature of 400°C, which corresponds to a wafer temperature of 235°C, with a chamber pressure of 50mTorr. Using the method of the invention, growing films thicker than 50Å will take more than 3 hours, approaching unreasonable lengths of time for the thickness.

15 A test of the nitride-like quality is the efficacy to block diffusion at high temperatures. This is demonstrated with the blocking of oxygen during a 1000°C dry oxidation for 18 minutes. A bare silicon wafer will oxidize to yield a 19 nm silicon dioxide layer. Approximately 1.2 nm of nitride will effectively block oxygen and less than 2.0 nm of oxide will form. This is illustrated in Fig. 3 for the room temperature and 400°C chuck temperature  
20 nitridation process. It is seen that the high temperature nitride is more effective as a diffusion block in the sub-1.0 nm region.

One sample nitrided with  $\text{N}_2$  rather than  $\text{NH}_3$  is illustrated in Figs 2 and 3. The  $\text{N}_2$

appears to hold promise as an alternative to  $\text{NH}_3$ .

With the advancement in excimer lamp technology, the use of alternate wavelengths are possible. Other excimers produce light at 126 nm, 146 nm, 222 nm, and 308 nm, however, these are probably not as efficiently as the  $\text{Xe}_2$  at 172 nm.

5               The configuration of the lamp can be altered to form a ring around the substrate or lie in a different orientation with respect to the substrate. A variety of lamp shapes are possible and can be oriented in a variety of positions to make this process work.

Thus, a method and system for low temperature nitridation of silicon has been disclosed. It will be appreciated that further variations and modifications thereof may be made  
10           within the scope of the invention as defined in the appended claims.